

# Package ‘PUPMSI’

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**Type** Package

**Title** Moisture Sorption Isotherm Modeling Program

**Version** 0.1.0

**Description** Contains sixteen moisture sorption isotherm models, which evaluate the fitness of adsorption and desorption curves for further understanding of the relationship between moisture content and water activity. Fitness evaluation is conducted through parameter estimation and error analysis. Moreover, graphical representation, hysteresis area estimation, and isotherm classification through the equation of Blahovec & Yanniotis (2009) <[doi:10.1016/j.jfoodeng.2008.08.007](https://doi.org/10.1016/j.jfoodeng.2008.08.007)> which is based on the classification system introduced by Brunauer et. al. (1940) <[doi:10.1021/ja01864a025](https://doi.org/10.1021/ja01864a025)> are also included for the visualization of models and hysteresis.

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**NeedsCompilation** no

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BETMSI

*Brunauer-Emmett-Teller(BET) Moisture Sorption Isotherm*

---

### Description

Brunauer-Emmett-Teller(BET) is a two-parameter isotherm model used for the optimum moisture content determination for drying and storage stability of foods, and in the food's surface area estimation.

### Usage

BETMSI(WaterAct, AdsorpM, DesorpM)

### Arguments

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

### Value

the nonlinear regression, parameters, and graphical visualization for the Brunauer-Emmett-Teller(BET) Moisture Sorption Isotherm model.

### Author(s)

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

## References

Aviara, N. A., et al. (2016). Effect of Temperature and Moisture Sorption Hysteresis on Monolayer Moisture Content of Selected Crops Determined Using BET and GAB Models. 37Th Annual Conference and Annual General Meeting-"Minna 2016," October.

Staudt, P. B., et al. (2013) <doi:10.1016/j.jfoodeng.2012.07.016> A new method for predicting sorption isotherms at different temperatures using the BET model. Journal of Food Engineering, 114(1), 139-145.

## Examples

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
BETMSI(WaterAct, AdsorpM, DesorpM)
```

---

BradleyMSI

*Bradley Moisture Sorption Isotherm*

---

## Description

Bradley model is a two-parameter isotherm model that measures polar nature of sorptive surfaces.

## Usage

```
BradleyMSI(WaterAct, AdsorpM, DesorpM)
```

## Arguments

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

## Value

the nonlinear regression, parameters, and graphical visualization for the Bradley Moisture Sorption Isotherm model.

## Author(s)

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

## References

Bradley, R. Stevenson (1936) <doi:10.1039/JR9360001467> Polymolecular adsorbed films. Part I. The adsorption of argon on salt crystals at low temperatures, and the determination of surface fields. Journal of the Chemical Society (Resumed), (), 1467-.

## Examples

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
BradleyMSI(WaterAct, AdsorpM, DesorpM)
```

---

CaurieMSI

*Caurie Moisture Sorption Isotherm*

---

## Description

Caurie model is a two-parameter isotherm created for calculation of water sorption data of dehydrated foods over a wide range of water activity.

## Usage

```
CaurieMSI(WaterAct, AdsorpM, DesorpM)
```

## Arguments

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

## Value

the nonlinear regression, parameters, and graphical visualization for the Caurie Moisture Sorption Isotherm model.

## Author(s)

Benz L. Rivera  
John Carlo F. Panganiban  
Kim M. Villacorte  
Chester C. Deocarís

## References

- Caurie, M. (1970) <doi:10.1111/j.1365-2621.1970.tb01571.x> A new model equation for predicting safe storage moisture levels for optimum stability of dehydrated foods. *International Journal of Food Science & Technology*, 5(3), 301-307.
- Caurie, M. (2007) <doi:10.1111/j.1365-2621.2006.01203.x> Hysteresis phenomenon in foods. *International Journal of Food Science and Technology*, 42(1), 45-49.
- Caurie, M. (2011) <doi:10.1007/978-90-481-3585-1\_71> Hysteresis in foods. In *Encyclopedia of Earth Sciences Series: Vol. Part 4* (p. 384). Springer Netherlands.

## Examples

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
CaurieMSI(WaterAct, AdsorpM, DesorpM)
```

GABMSI

*Guggenheim-Anderson-de Boer(GAB) Moisture Sorption Isotherm*

## Description

GAB model is a multimolecular, localized and homogeneous adsorption model, is one of the most versatile models considering multilayer adsorption at high water activity values.

## Usage

```
GABMSI(WaterAct, AdsorpM, DesorpM)
```

## Arguments

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

## Value

the nonlinear regression, parameters, and graphical visualization for the Guggenheim-Anderson-de Boer(GAB) Moisture Sorption Isotherm model.

## Author(s)

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

## References

Aviara, N. A. (2020) <doi:10.5772/intechopen.87996> Moisture Sorption Isotherms and Isotherm Model Performance Evaluation for Food and Agricultural Products. In Sorption in 2020s. IntechOpen.

Aviara, N. A., et al. (2016). Effect of Temperature and Moisture Sorption Hysteresis on Monolayer Moisture Content of Selected Crops Determined Using BET and GAB Models. 37Th Annual Conference and Annual General Meeting-"Minna 2016," October.

Maroulis, Z. B., et al. (1988) <doi:10.1016/0260-8774(88)90069-6> Application of the GAB model to the moisture sorption isotherms for dried fruits. Journal of Food Engineering, 7(1), 63-78.

Prasantha, B. D. R. (2018). Prediction of Moisture Adsorption Characteristics of Dehydrated Fruits Using the GAB. Ann Agric Crop Sci, 3(1), 1-4.

## Examples

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
GABMSI(WaterAct, AdsorpM, DesorpM)
```

---

HailHorroMSI

*Hailwood-Horrobin (HH) Moisture Sorption Isotherm*

---

## Description

Hailwood-Horrobin (HH) model is an example of multilayer surface sorption model, is suitable for analysis of experimental wood moisture sorption (WMS) isotherms.

## Usage

```
HailHorroMSI(WaterAct, AdsorpM, DesorpM)
```

## Arguments

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

## Value

the nonlinear regression, parameters, and graphical visualization for the Hailwood-Horrobin (HH) Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Hailwood, A. J., & Horrobin, S. (1946) <doi:10.1039/TF946420B084> Absorption of water by polymers: Analysis in terms of a simple model. Transactions of the Faraday Society, 42(0), B084-B092.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
HailHorroMSI(WaterAct, AdsorpM, DesorpM)
```

---

HalseyMSI

*Halsey Moisture Sorption Isotherm*


---

**Description**

Halsey Isotherm is a 2-parameter model which expresses condensation of multilayers at proportionally large distances from the surface considering the assumption that a molecule's potential energy is inversely proportional to the nth power of its distance from the surface.

**Usage**

```
HalseyMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Halsey Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Halsey, G. (1948) <doi:10.1063/1.1746689> Physical adsorption on non-uniform surfaces. The Journal of Chemical Physics, 16(10), 931-937.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
HalseyMSI(WaterAct, AdsorpM, DesorpM)
```

---

 HendersonMSI

---

*Henderson Moisture Sorption Isotherm*


---

**Description**

Henderson Isotherm is an empirical two-parameter equation for moisture adsorption of food products, useful in predicting moisture content for different water activity levels.

**Usage**

```
HendersonMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Henderson Moisture Sorption Isotherm model.



**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Stencl, J. (2004) <doi:10.1260/0263617042863039> Moisture Sorption Isotherms of Whey Powder Spray in the 10-40C Temperature Range. *Adsorption Science & Technology*, 22(5), 377-384.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
HendersonMSI(WaterAct, AdsorpM, DesorpM)
```

---

HysteresisMSI

*Hysteresis Area, Brunauer Classification System*


---

**Description**

Hysteresis area evaluation via trapezoidal approximation.

**Usage**

```
HysteresisMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the measurement of hysteresis, classification of isotherms, and graphical visualization for the observed values of moisture sorption isotherms.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Caurie, M. (2007) <doi:10.1111/j.1365-2621.2006.01203.x> Hysteresis phenomenon in foods. International Journal of Food Science and Technology, 42(1), 45-49.

Brunauer, S., et al. (1940) <doi:10.1021/ja01864a025> On a Theory of the van der Waals Adsorption of Gases. Journal of the American Chemical Society, 62(7), 1723-1732.

Blahovec J., & Yanniotis S. (2009) <doi:10.1016/j.jfoodeng.2008.08.007> Modified classification of sorption isotherms. J Food Eng. 2009 Mar; 91 (1): 72-77

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
HysteresisMSI(WaterAct, AdsorpM, DesorpM)
```

---

 IgleChiMSI

---

*Iglesias-Chirife Moisture Sorption Isotherm*


---

**Description**

Iglesias-Chirife Isotherm is an empirical equation for describing water sorption behavior of various fruits and other sugar-rich foods.

**Usage**

```
IgleChiMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Iglesias-Chirife Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Chirife, J., & Iglesias, H. A. (1978) <doi:10.1111/j.1365-2621.1978.tb00792.x> Equations for fitting water sorption isotherms of foods: Part 1 - A review. *International Journal of Food Science & Technology*, 13(3), 159-174.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
IgleChiMSI(WaterAct, AdsorpM, DesorpM)
```

---

 KuhnMSI

---

*Kuhn Moisture Sorption Isotherm*


---

**Description**

Kuhn Isotherm is a two-parameter model which contains many defining characteristics wherein each surface site has a different adsorption potential, as well as cluster formations on each site due to increase in partial pressure.

**Usage**

```
KuhnMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Kuhn Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Bi, Y., et al. (2018). The prediction of moisture adsorption isotherm for commercial sodium bicarbonate powder. *International Journal of Scientific & Engineering Research*, 9(3).

Kuhn, I. (1967) <doi:10.1016/0021-9797(67)90202-0> A generalized potential theory of adsorption. I. The derivation of a general equation for adsorption isotherms. *Journal of Colloid And Interface Science*, 23(4), 563-571.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
KuhnMSI(WaterAct, AdsorpM, DesorpM)
```

---

 LangmuirMSI

---

*Langmuir Moisture Sorption Isotherm*


---

**Description**

Langmuir Isotherm is a two-parameter model applicable for unimolecular layers with similar sorption sites. Langmuir's isotherm is the most crucial equation among the theoretical models, whose basis are the forces acting between the product surface and the condensed water from the vapor as a monomolecular layer.

**Usage**

```
LangmuirMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Langmuir Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Andrade, R. D., et al. (2011). Models of sorption isotherms for food: Uses and limitations. *Vitae*. In *Vitae* (Vol. 18, Issue 3). Facultad De Qui??mica Farmace??utica, Universidad de Antioquia. [http://www.scielo.org.co/scielo.php?script=sci\\_arttext&pid=S0121-40042011000300012&lng=en&nrm=iso&tlng=en](http://www.scielo.org.co/scielo.php?script=sci_arttext&pid=S0121-40042011000300012&lng=en&nrm=iso&tlng=en)

Saroyda, J. V., Cruz, et al. (2020) <doi:10.1016/S0001-8686(00)00082> Package "PUPAIM" Type Package Title A Collection of Physical and Chemical Adsorption Isotherm Models Version 0.2.0. <doi:10.1016/S0001-8686(00)00082>

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
LangmuirMSI(WaterAct, AdsorpM, DesorpM)
```

---

 Lewicki2MSI

---

*Lewicki-2-Parameter Moisture Sorption Isotherm*


---

**Description**

Lewicki-2-Parameter MSI is a two-parameter sorption model that was developed based on Raoult's law, which assumes that water is present either as free water or as water of hydration.

**Usage**

```
Lewicki2MSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for Lewicki-2-Parameter model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Lewicki, P. P. (2000) <doi:10.1016/S0260-8774(99)00130-2> Raoult's law based food water sorption isotherm. *Journal of Food Engineering*, 43(1), 31-40.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
Lewicki2MSI(WaterAct, AdsorpM, DesorpM)
```

---

 Lewicki3MSI

---

*Lewicki-3-Parameter Moisture Sorption Isotherm*


---

**Description**

The three-parameter Lewicki model is most suitable for describing the sorption characteristics of raw potato, potato starch, starch-sugar and starch-salt gels within specific temperature and water activity ranges.

**Usage**

```
Lewicki3MSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for Lewicki-3-Parameter model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

McMinn, W. A., et al. (2004) <doi:10.1002/jsfa.1866> Assessment of two- and three-parameter Lewicki models for description of sorption phenomena of starch materials. *Journal of the Science of Food and Agriculture*, 84(13), 1695-1700.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
Lewicki3MSI(WaterAct, AdsorpM, DesorpM)
```

---

 ModChenMSI

---

*Modified Chen Moisture Sorption Isotherm*


---

**Description**

Modified Chen is 2-parameter model related to the drying principle. It is restricted to situations where diffusion is the primary mode of mass transport and is focused on the steady state of the drying equation.

**Usage**

```
ModChenMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Modified Chen Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

- Chen, C. (2019) <doi:10.3390/foods8060191> Validation of the Component Model for Prediction of Moisture Sorption Isotherms of Two Herbs and other Products. *Foods*, 8(6), 191.
- Chen, C. S. (1971) <doi:10.13031/2013.38421> Equilibrium Moisture Curves for Biological Materials. *Transactions of the ASAE*, 14(5), 0924-0926.
- Chen, C. S. & Clayton, J. T. (1971) <doi:10.13031/2013.38422> The Effect Of Temperature On Sorption Isotherms Of Biological Materials. *Transactions of the ASAE*, 14(5), 0927-0929.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
ModChenMSI(WaterAct, AdsorpM, DesorpM)
```

---

 OswinMSI

---

*Oswin Moisture Sorption Isotherm*


---

**Description**

An empirical model developed through a series of mathematical equations that consists in a series expansion for sigmoidal curves.

**Usage**

```
OswinMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Oswin Moisture Sorption Isotherm model.



**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Oswin, C. R. (1946) <doi:10.1002/jctb.5000651216> The kinetics of package life. III. The isotherm. Journal of the Society of Chemical Industry, 65(12), 419-421.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
OswinMSI(WaterAct, AdsorpM, DesorpM)
```

---

 PelegMSI

---

*Peleg Moisture Sorption Isotherm*


---

**Description**

Peleg model is an empirical 4-parameter isotherm which describes sigmoidal and non-sigmoidal behavior of isotherm plots.

**Usage**

```
PelegMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Peleg Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
 John Carlo F. Panganiban  
 Kim M. Villacorte  
 Chester C. Deocaris

**References**

Abu-Ghannam, N., & McKenna, B. (1997) <doi:10.1016/S0260-8774(97)00034-4> The application of Peleg's equation to model water absorption during the soaking of red kidney beans (*Phaseolus vulgaris* L.). *Journal of Food Engineering*, 32(4), 391-401.

Peleg, M. (1993) <doi:10.1111/j.1745-4530.1993.tb00160.x> Assessment of a semi-empirical four parameter general model for sigmoid moisture sorption isotherms. *Journal of Food Process Engineering*, 16(1), 21-37.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
PelegMSI(WaterAct, AdsorpM, DesorpM)
```

---

 SmithMSI

---

*Smith Moisture Sorption Isotherm*


---

**Description**

Smith MSI is an empirical isotherm model for curve assessment of water sorption of polymers with high molar mass.

**Usage**

```
SmithMSI(WaterAct, AdsorpM, DesorpM)
```

**Arguments**

WaterAct	the numerical value of Water Activity, which ranges from 0 to 1.
AdsorpM	the numerical value of the Moisture content of the Adsorption curve, which ranges from 0 to 1.
DesorpM	the numerical value of the Moisture content of the Desorption curve, which ranges from 0 to 1.

**Value**

the nonlinear regression, parameters, and graphical visualization for the Smith Moisture Sorption Isotherm model.

**Author(s)**

Benz L. Rivera  
John Carlo F. Panganiban  
Kim M. Villacorte  
Chester C. Deocaris

**References**

Smith, S. E. (1947) <doi:10.1021/ja01195a053> The Sorption of Water Vapor by High Polymers. Journal of the American Chemical Society, 69(3), 646-651.

**Examples**

```
WaterAct <- c(0.1145,0.2274,0.3265,0.4291,0.6342,0.7385,0.8274,0.9573)
AdsorpM <- c(0.0234, 0.0366, 0.0496, 0.0648, 0.0887, 0.1096, 0.1343, 0.1938)
DesorpM <- c(0.0459, 0.0637, 0.0794, 0.0884, 0.1158, 0.1298,0.1500, 0.1938)
SmithMSI(WaterAct, AdsorpM, DesorpM)
```

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